

Assignment Temperature Conversions and Thermal Energy

Due May 6, 2020

Total: [/16]

Name:

- Convert the following temperatures to the indicated scales.
 - $78 \text{ F} = \underline{\hspace{2cm}} \text{ } ^\circ\text{C}$
 - $67 \text{ } ^\circ\text{C} = \underline{\hspace{2cm}} \text{ K}$
 - $42 \text{ } ^\circ\text{C} = \underline{\hspace{2cm}} \text{ F}$
 - $100 \text{ F} = \underline{\hspace{2cm}} \text{ K}$ [/4]
- What mass of silver, in grams, goes from -16°C to 58°C when heated by 11.7 kJ of heat? [/4]
- Find the initial temperature of 0.355 kg of glycerin if it is heated by 600 J of heat to produce a final temperature of 91°C .? [4]
- Brooklyn heats 10 kg of an unknown substance from 10°C to 40°C and finds that it produces $152\,100 \text{ J}$ of thermal energy. Determine the specific heat capacity, C , of the substance and using the chart determine what the substance is. [/4]